

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
10 October 2002 (10.10.2002)

PCT

(10) International Publication Number  
**WO 02/078584 A1**

(51) International Patent Classification<sup>7</sup>: **A61F 7/08**  
(21) International Application Number: **PCT/IN01/00074**  
(22) International Filing Date: **30 March 2001 (30.03.2001)**  
(25) Filing Language: **English**  
(26) Publication Language: **English**  
(71) Applicant: **COUNCIL OF SCIENTIFIC AND INDUSTRIAL RESEARCH [IN/IN];** Rafi Marg, New Delhi 110 001 (IN).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

(72) Inventors: **SHARMA, Chhatrapal**; National Physical Laboratory, K.S Krishnan Marg, New Delhi 110 012 (IN). **SHARMA, Rajendra, Kumar**; National Physical Laboratory, K.S Krishnan Marg, New Delhi 110 012 (IN). **KANT, Chandra**; National Physical Laboratory, K.S Krishnan Marg, New Delhi 110 012 (IN). **SARKAR, Ajit, Kumar**; National Physical Laboratory, K.S Krishnan Marg, New Delhi 110 012 (IN).

**Published:**

— *with international search report*

*For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.*

(74) Agents: **SUBRAMANIAM, Hariharan** et al.; Subramaniam, Nataraj & Associates, E- 556, Greater Kailash - II, New Delhi 110 048 (IN).



**WO 02/078584 A1**

(54) Title: **REUSABLE HEAT PACK**

(57) Abstract: The invention discloses a reusable heat pack comprising a soft sealed plastic bag containing a mixture of supercooled supersaturated recrystallizable solution of inorganic substance, alcohol, chemical compound additives, and means to trigger recrystallization of supercooled supersaturated recrystallizable solution of inorganic substance.

## **REUSABLE HEAT PACK**

### **Field of the invention**

The present invention relates to a reusable heat pack. The present invention  
5 also relates to a trigger device for a heat pack. The invention also relates to a  
supercooled supersaturated recrystallizable solution for use in heat pack and to a  
process for the preparation thereof.

### **Background of the invention**

Use of heating or cooling for many applications has been done by a variety of  
10 techniques depending on the specific need. For example, persons suffering from  
muscular pains normally use rubber bags filled with hot boiling water to alleviate the  
pain. In case of any head injury normally the doctor's prescription is to apply ice as a  
first aid. In case of high fever the common house remedy has been to apply a cloth  
soaked in cold water on the forehead to bring down the temperature. Many a time on a  
15 recreational trip people tend to take packed food or even carry cooking apparatus and  
the raw, materials for cooking food. The other need of heat energy is in the remote  
areas where availability of the fuels may not be possible or may be very expensive.  
Therefore to get over all these difficult situations it is necessary to devise ways and  
means to generate heat in an economical way and preferably without the use of  
20 conventional fuels. To address this very need, development of device called the heat  
pack has occurred over the past 15 years. The main principle underlying the heat pack  
is to use heat generated due to a start of a chemical reaction between two or more  
chemicals or due to triggering of initiation of crystallization of a supercooled  
chemical solution. In the context of the present invention it is the second aspect which  
25 is disclosed

Several warming materials are available in the market such as hot water bag,  
portable electrical bag warmer or battery operated warmer. However, in case of hot  
water bag, hot water is to be poured in the bag which can only be obtained either by  
using fuel or electrical energy. Further the heat of the water bag does not last long  
30 enough for giving any observable effect. In case of electrical bag warmer a permanent  
supply of electricity is needed. This limits the use of this heating device to only such  
places where there is electricity available i.e only indoors. In case of battery operated  
warmer the battery needs to be charged regularly and needs replacement as and when

it gets exhausted. In all these cases another drawback is that the use is very limited and safety precautions need to be taken in case of electrical heat packs.

A non-conventional heat pack has been disclosed in a US Patent 4,532,110 (1985). In this disclosure heat energy is generated by a trigger action on a supercooled saturated solution of sodium acetate. The triggering is accomplished by a specially designed device. The entire assembly of trigger and the supercooled solution is kept in a plastic container made of PVC, polyurethane or the like. The drawback in the invention is that the trigger used to initiate the crystallization of the super cooled sodium acetate solution has to have a protective layer of gold or silver to prevent any corrosion due to contact with the chemical. This is not desirable since the material used for the trigger is already an expensive CuBe alloy. Further the PVC material for the heat pack manufacturing has a very limited shelf life due to inevitable leaching of the chloride ion due to the presence of sodium salt solution. This drawback is present in the other disclosed inventions as well as disclosed in US Patents 4,572,158; 5,058,563 and 5,205,278. A further drawback in the disclosed inventions, except in 5,058,563, is that the supercooled solution tends to flow to one side giving rise to uneven heating effects thereby rendering the heat pack difficult to use by needing to adjust the contour of the pack for a particular application before initiation of crystallisation and resultant heat generation.

In US Patent 5,058,563 use has been made of an additive to induce gelation during crystallisation. This gelation helps in maintaining the desired density of the solution and hence uniformity of heat over the entire useful area. The gelation material used in the said disclosure is a water soluble hydroethyl cellulose polymer. This material though giving the advantage of gelling the solution suffers from the disadvantage of being unstable, thereby rendering the heat pack incapable of reuse.

US Patent 5,843,145 discloses a reusable heat pack wherein a mixture of three constituents forming a gel, is sealed in a plastic bag. This reusable heat pack needs an external heating or cooling means to give a useful application of the device. In case of heating application, the heat pack is put in a microwave oven or any other heating means to energise the pack. It is claimed that the temperature of the heat pack is maintained for a sufficient length of time thereby claiming its use as a heat pack. In view of the fact that the pack needs an external heating means, the invention proves to be disadvantageous in its claimed use as a heat pack. This is mainly due to the fact that in the invention discloses that the preconditioning of the pack is to be done for a

time period typically two hours. This renders the pack as disclosed not advantageous in an emergency practical situation and also in remote areas, where heating means such as microwave oven may not be available.

JP 58180575 discloses a cooling pack with a mixture provided therein  
5 containing inter alia, aniline chloride. However, in this disclosure, the mixture acts as an endothermic material wherein the material when dissolved in water absorbs heat and produces a cooling effect.

The present invention circumvents the drawbacks of the inventions of the prior art.

#### 10 **Objects of the invention**

An object of the present invention is to provide a reusable heat pack which obviates the drawbacks mentioned above.

Another object of the invention is to provide a heat pack with a shelf life of at least six months.

15 A further objective of the present invention is to provide a heat pack which can be recycled at least a hundred times.

Still another objective of the present invention is to provide a heat pack with heat retention for at least two hours.

Yet another object of the present invention is to provide a heat pack capable of  
20 being used for therapeutic application.

#### **Summary of the invention**

Accordingly the present invention provides a reusable heat pack, which comprises a soft sealed plastic bag containing a mixture of supercooled supersaturated recrystallizable solution of inorganic substance, alcohol, chemical compound  
25 additives, and means to trigger recrystallization of supercooled supersaturated recrystallizable solution of inorganic substance.

In an embodiment of the invention the plastic bag is made of polypropylene or polyvinyl chloride.

In another embodiment of the invention, the inorganic substance is selected  
30 from sodium acetate and calcium nitrate.

In still another embodiment of the present invention the supercooled supersaturated recrystallizable solution of inorganic substance is made in good quality water.

In a further embodiment of the invention, the good quality water is bacteria free municipal supply or distilled water.

In a further embodiment of the present invention the alcohol is selected from ethylene glycol and glycerol.

5 In yet another embodiment of the present invention the chemical additive comprises an organic amine.

In a further embodiment of the invention, the organic amine comprises aniline.

In another embodiment of the invention, the supercooled supersaturated recrystallizable solution comprises a mixture of hydrated salt of sodium acetate 50 –  
10 70 wt%, ethylene glycol 1 – 10 wt%, aniline 10 – 20 wt% and water 10 – 29 wt%.

In another embodiment of the present invention the means for triggering recrystallization comprises a mechanical device selected from the group consisting of crocodile clip and a safety pin.

In a further embodiment of the present invention the material of the means for  
15 trigger comprises a noncorrosive material.

In a further embodiment of the invention, the non-corrosive material for the trigger means comprises stainless steel.

In a further embodiment of the invention the heat pack has a shelf life of at least six months.

20 In still another embodiment of the invention the heat pack is capable of being recycled for at least 100 cycles.

In another embodiment of the present invention the heat pack retains heat for at least two hours.

In still another embodiment of the present invention the heat pack is used for  
25 therapeutic applications, body warming in high altitudes.

The invention also relates to a process for the preparation of a reusable heat pack said process comprising preparing a hydrated solution of an inorganic substance in water, adding an alcohol to this solution and mixing thoroughly, adding aniline and thoroughly mixing the resultant mixture, enclosing the mixture in a soft sealed plastic  
30 bag and adding a trigger means thereto.

The present invention also relates to a mixture for use in reusable heat packs, said mixture comprising a solution of an inorganic substance in water, alcohol and aniline.

In one embodiment of the invention, the mixture comprises a mixture of hydrated salt of sodium acetate 50 – 70 wt%, ethylene glycol 1 – 10 wt%, aniline 10 – 20 wt% and water 10 – 29 wt%.

The invention also relates to process for the preparation of a mixture for use in reusable heat packs, said process comprising preparing a hydrated solution of an inorganic substance in water, adding an alcohol to this solution and mixing thoroughly, adding aniline and thoroughly mixing the resultant mixture.

The mixture of the invention for use in a reusable heat pack is a synergistic admixture with properties over and above the aggregation of properties of the individual constituents and is not a simple admixture.

#### **Detailed description of the invention**

First a hydrated solution of preferably sodium acetate of a preferred purity of was prepared in ordinary water by mixing 500-700 gms. in 0.1 to 0.3 liters of water. The solution was prepared by thoroughly mixing sodium acetate in water. To this was added a polyhydric alcohol and preferably ethylene glycol. This mixture was thoroughly mixed and to this was then added another additive and preferably aniline and the resultant mixture was thoroughly mixed. The addition of polyhydric alcohol is to provide smoothness to the recrystallised supercooled saturated solution of sodium acetate. This is achieved by controlled increase in viscosity of the resulting mixture after recrystallisation due to triggering action by a mechanical device. The use of aniline in the mixture of the heat pack increases the shelf life of the heat pack and also to give the unique property of recyclability of the heat pack. After the mixture has been prepared the same is poured into a plastic bag of preferred dimension of 100mm<sup>2</sup> and made of thin sheet of polyvinyl chloride (PVC). The use of aniline as an additive gives the desired property of preventing the leaching of the plastic material and thereby enhancing the shelf life. A stainless steel crocodile clip is inserted in the heat pack and then finally sealed from the open end. The heat pack is now ready having typical contents, but not limited to, of hydrated salt of sodium acetate 50wt%, ethylene glycol 1wt%, aniline 20wt% and water 29 wt%. The recrystallisation activity was initiated by softly pressing the crocodile clip and instantaneously releasing the same. The recrystallization took place and instantaneously heat was generated over the entire heat pack. This heat pack could be easily contoured over a body part and the part warmed thereby. The temperature of the pack was 45°C above ambient and maintained for about two to three hours depending on the size of the heat pack.

In a typical therapeutic application, after energizing the pack the same is wrapped in a towel and applied on the patient. The cooling of the pack to ambient took place in about four hours time. The pack was reactivated by putting the same in hot water resulting in melting of the crystallized content of the pack and then taking it out from hot water and cooled to ambient temperature. This way the heat pack was ready for use again.

The principle underlying the invention is that the formation of the hydrated salt results in an endothermic reaction absorbing heat of crystallization. When the crystallization is effected by the triggering action the heat of crystallization is released to the container thereby heating the exterior surface of the container. This renders the device useful for any application. The temperature attainable with the heat pack can be fixed by the amount of the supercooled supersaturated solution of the hydrated salt. Accordingly the shelf life of the device will also be adjusted by correspondingly increasing the quantity of the aniline or any of the other additives like hydroxides or carbonates.

The novelty of the present invention lies in the heat pack being reusable and having a long shelf life and being capable of recyclable.

This novelty of the heat pack is due to the inventive step of adding hydroxides, carbonates or amines to the recrystallizable mixture.

In the embodiment of the examples the heat pack is manufactured by first dissolving hydrated salt of sodium acetate in water. Ethylene glycol was added to this. This was mixed thoroughly and to this aniline was added. Again the mixture was thoroughly mixed and then poured in a plastic bag of PVC. The trigger used is a stainless steel crocodile clip and is put inside the bag and then the whole assembly was sealed.

The following examples are given by way of illustration only and should not be construed to limit the scope of the invention.

#### **Example 1:**

Heat pack contained hydrated salt of sodium acetate 50 wt%, ethylene glycol 1 wt%, aniline 20 wt% and water 29 wt%. Triggering action was done by pressing the clip three times. The crystallization resulted in instantaneous temperature rise to 75°C. The cooling of the pack to reach to ambient temperature took place in about four hours time.

**Example 2:**

Heat pack contained hydrated salt of sodium acetate 60 wt%, ethylene glycol 4wt%, aniline 15 wt% and water 21 wt%. The triggering action was done by pressing the clip three times. The crystallization resulted in instantaneous temperature rise to 65°C. The cooling of the pack to reach to ambient temperature took place in about four hours time.

**Example 3:**

Heat pack contained hydrated salt of sodium acetate 70 wt%, ethylene glycol 6wt%, aniline 10 wt% and water 14 wt%. The triggering action was done by pressing the clip three times. The crystallization resulted in instantaneous temperature rise to 75°C. The cooling of the pack to reach to ambient temperature took place in about four hours time.

**Example 4:**

Heat pack contained hydrated salt of sodium acetate 70 wt%, ethylene glycol 10wt%, aniline 10 wt% and water 10 wt%. The triggering action was done by pressing the clip three times. The crystallization resulted in instantaneous temperature rise to 70°C. The cooling of the pack to reach to ambient temperature took place in about four hours time.

Table 1 shows the consolidated results of the examples.

**Table -1**

Item	Weight percent of materials			
Sodium acetate	50	60	70	70
Ethylene Glycol	1	4	6	10
Aniline	20	15	10	10
Water	29	21	14	10
Temp of Heat pack °C	75	65	75	70

**The main advantages of the present invention are:**

1. It is easy to manufacture from commonly available chemicals and of moderate purity.
2. The heat pack can be stored for at least six months



3. The pack can be easily put on any contour due to its being flexible even after re crystallization.

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**We claim:**

1. A reusable heat pack, which comprises a soft sealed plastic bag containing a mixture of supercooled supersaturated recrystalizable solution of inorganic substance, alcohol, aniline, and means to trigger recrystallization of supercooled  
5 supersaturated recrystalizable solution of inorganic substance.
2. A reusable heat pack as claimed in claim 1 wherein the plastic bag is made of polypropylene or polyvinyl chloride.
3. A reusable heat pack as claimed in claim 1 wherein the inorganic substance is selected from sodium acetate and calcium nitrate.
- 10 4. A reusable heat pack as claimed in claim 1 wherein the supercooled supersaturated recrystalizable solution of inorganic substance is made in substantially pure water.
5. A reusable heat pack as claimed in claim 4 wherein the substantially pure water comprises bacteria free municipal supply or distilled water.
- 15 6. A reusable heat pack as claimed in claim 1 wherein the alcohol used is selected from ethylene glycol and glycerol.
7. A reusable heat pack as claimed in claim 1 wherein the supercooled supersaturated recrystalizable solution comprises a mixture of hydrated salt of sodium acetate 50 – 70 wt%, ethylene glycol 1 – 10 wt%, aniline 10 – 20 wt% and  
20 water 10 – 29 wt%.
8. A reusable heat pack as claimed in claim 1 wherein the means for triggering recrystallization comprises a mechanical device selected from the group consisting of crocodile clip and a safety pin.
9. A reusable heat pack as claimed in claim 1 wherein the material of the means for  
25 trigger comprises a noncorrosive material.
10. A reusable heat pack as claimed in claim 11 wherein the non-corrosive material for the trigger means comprises stainless steel.
11. A supercooled supersaturated recrystalizable solution for use in a heat pack, said supercooled super saturated recrystallisable solution comprises a mixture of a  
30 solution of an inorganic substance, alcohol and aniline.
12. A supercooled supersaturated recrystalizable solution as claimed in claim 11 comprising a mixture of hydrated salt of sodium acetate 50 – 70 wt%, ethylene glycol 1 – 10 wt%, aniline 10 – 20 wt% and water 10 – 29 wt%.

13. A process for the the preparation of a mixture for use in reusable heat packs, said process comprising preparing a hydrated solution of an inorganic substance in water, adding an alcohol to this solution and mixing thoroughly, adding aniline and thoroughly mixing the resultant mixture.

- 5 14. A method for the manufacture of a reusable heat pack said process comprising preparing a hydrated solution of an inorganic substance in water, adding an alcohol to this solution and mixing thoroughly, adding aniline and thoroughly mixing the resultant mixture, enclosing the mixture in a soft sealed plastic bag and adding a trigger means thereto.

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# INTERNATIONAL SEARCH REPORT

International Application No

PCI/IN 01/00074

**A. CLASSIFICATION OF SUBJECT MATTER**  
IPC 7 A61F7/08

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)  
IPC 7 A61F C09K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	GB 2 055 054 A (KAY LABORATORIES INC) 25 February 1981 (1981-02-25) page 3, line 39 - line 50; table 1 page 4, line 47	1,11,13, 14
A	US 3 913 559 A (DANDLIKER WALTER B) 21 October 1975 (1975-10-21) column 6, line 46 - line 59; table 1	1,11,13, 14
A	US 5 205 278 A (WANG CHING-CHUAN) 27 April 1993 (1993-04-27) cited in the application column 3, line 45 -column 3, line 53; claim 1	1
A	US 5 736 110 A (ANGELILLO STEPHEN P ET AL) 7 April 1998 (1998-04-07) column 7, line 58 - line 66	1,9,10

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents :

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Date of the actual completion of the international search

28 August 2001

Date of mailing of the international search report

04/09/2001

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Mayer, E

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/IN 01/00074

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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US 3913559	A	21-10-1975	AU 5191473 A 08-08-1974 CA 1015709 A 16-08-1977 DE 2305504 A 16-08-1973 FR 2171135 A 21-09-1973 GB 1426065 A 25-02-1976 IT 977209 B 10-09-1974 JP 48091641 A 28-11-1973
US 5205278	A	27-04-1993	NONE
US 5736110	A	07-04-1998	NONE

DOCKET NO: WSO-41953

SERIAL NO: 10/629,926

APPLICANT: Gluderer

LERNER AND GREENBERG P.A.

P.O. BOX 2480

HOLLYWOOD, FLORIDA 33022

TEL. (954) 925-1100